

Polymer/ceramic Composite Electrolyte for Solid State Batteries

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Safety



How Lithium-Ion Batteries Turn Into Skin-Searing Firebombs

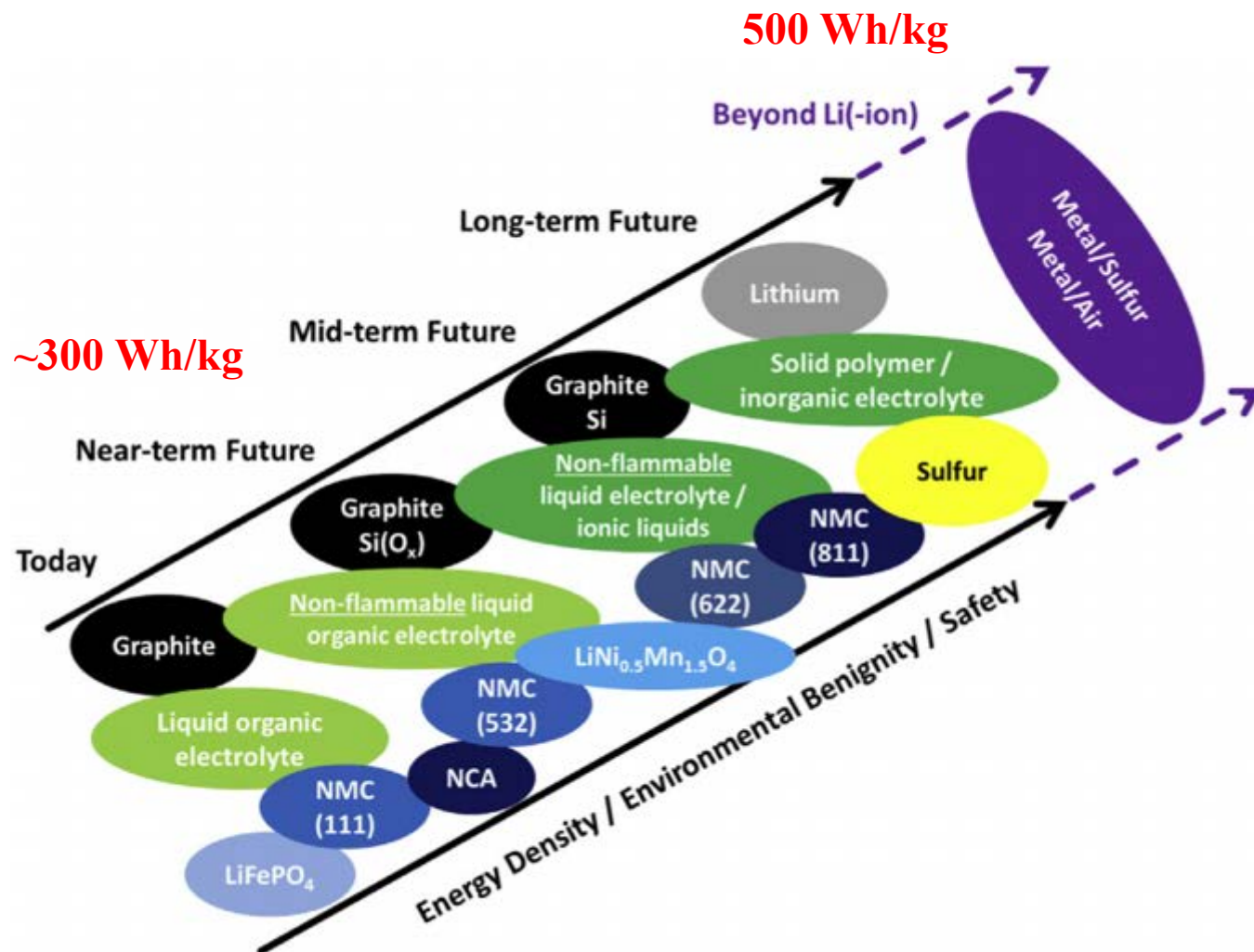
Burning concern: Energy storage industry battles battery fires

Morning Mix

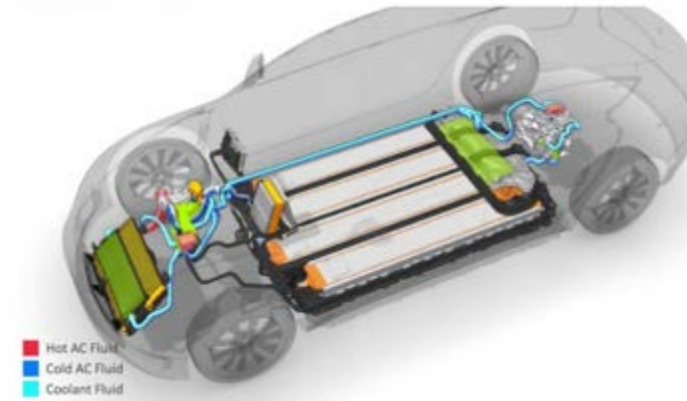
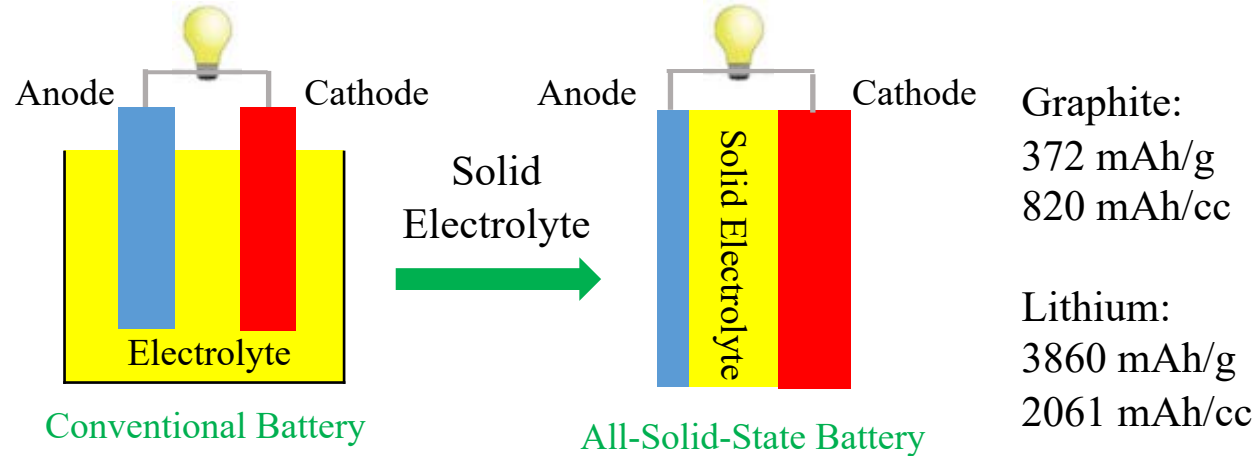
Woman's headphone batteries exploded on flight to Australia, causing burns

Hyperdrive

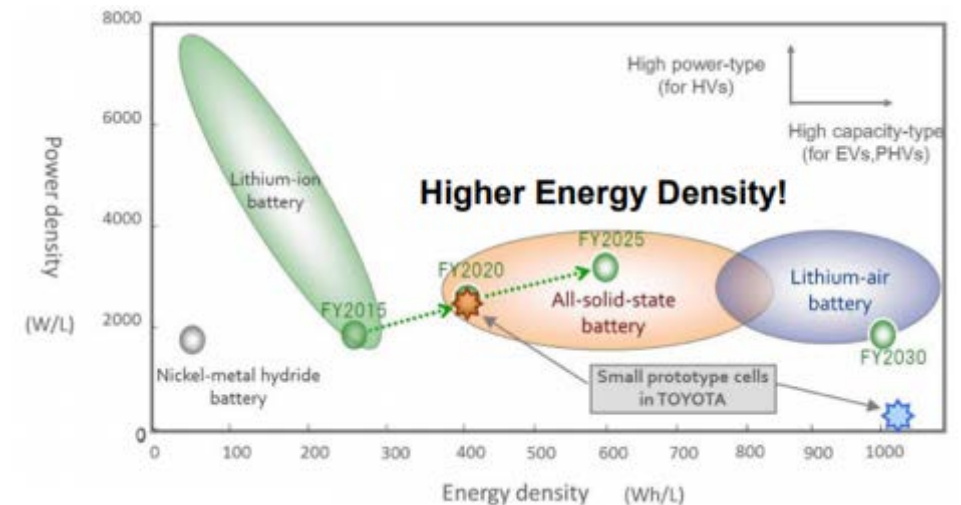
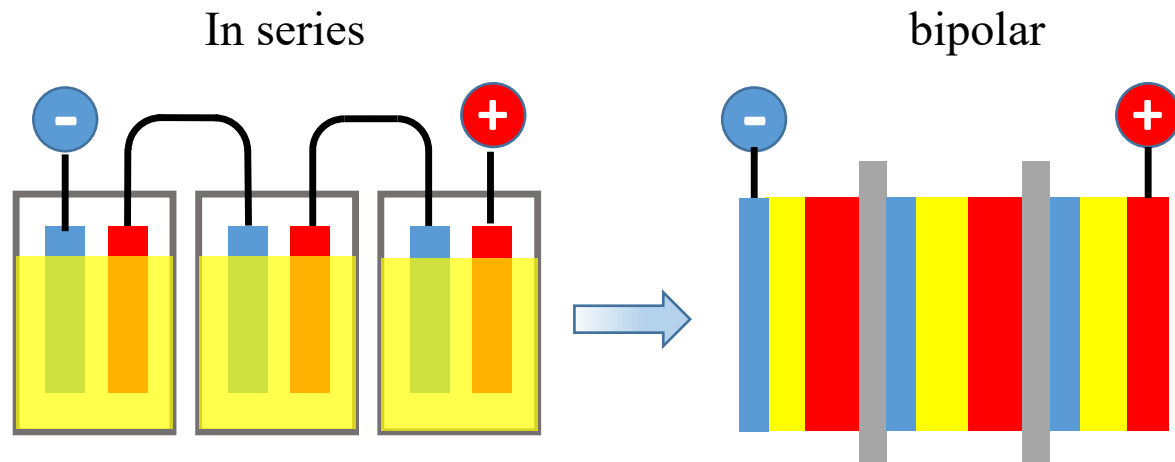
Explosions Threatening Lithium-Ion's Edge in a Battery Race



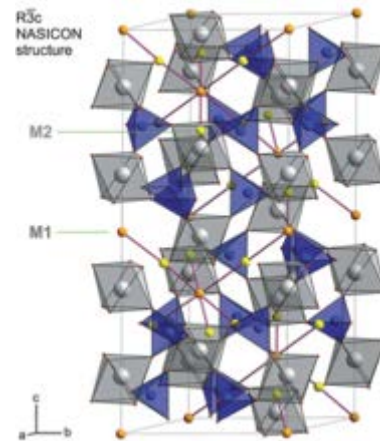
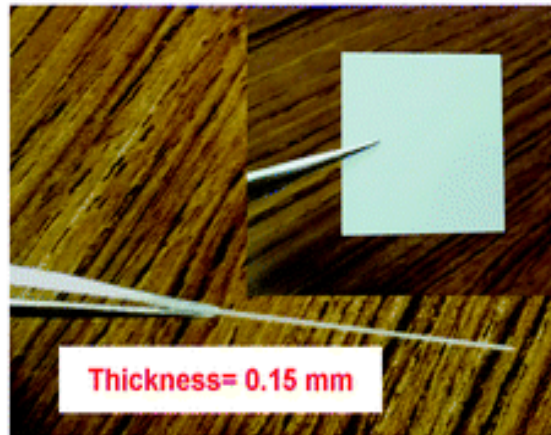
Energy Density of Solid-state Batteries



- Simplified thermal management system

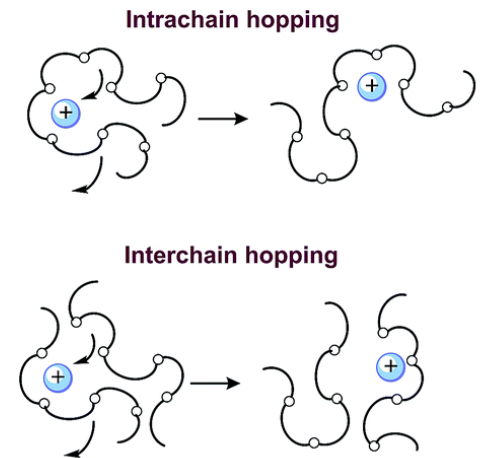


Ceramic Electrolyte



$\text{Li}_{1+x}\text{Al}_x\text{Ge}_{2-x}(\text{PO}_4)_3$ (LAGP),
 $\text{Li}_7\text{La}_3\text{Zr}_2\text{O}_{12}$ (LLZO),
 $\text{Li}_7\text{P}_3\text{S}_{11}$ (LPS)
 $\text{Li}_{10}\text{GeP}_2\text{S}_{12}$ (LGPS)

Polymer Electrolyte



Poly(ethylene oxide)
 Polycarbonates (e.g.
 polyethylene carbonate)
 Poly(FEC)
 Polyacrylonitrile (PAN).....

Solid Electrolyte Classes

	Conductivity (S/cm)	Cathode Stability	Anode Stability	Processing / Scalability	Air Stability
Sulfides (e.g. $\text{Li}_7\text{P}_3\text{S}_{11}$)	$10^{-3} - 10^{-2}$ at 25 °C				
Oxides (e.g. LLZO)	$10^{-4} - 10^{-3}$ at 25 °C				
Polymer (e.g. PEO, PAN)	$10^{-5} - 10^{-4}$ at 25 °C, 10^{-3} at 60 °C				



Good



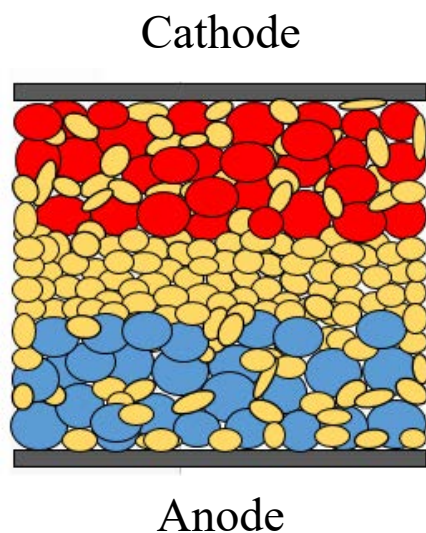
OK



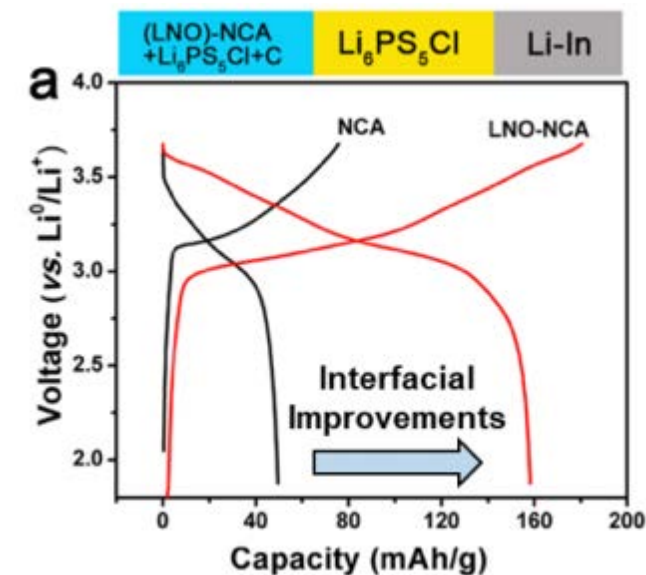
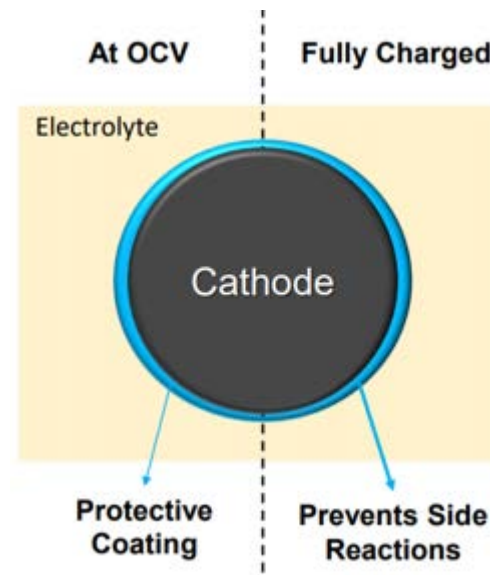
Not good

Recent Progress

Toyota
NMC/sulfide/graphite



High power
Wide temperature range



Anode stability: Lithium $\rightarrow$ Graphite or Indium

Potential Challenges:

Scalability, Air stability, Pressure (5-20 MPa)?

Recent Progress

Bolloré blue car:
LiFePO₄/PEO/Li metal batteries



Challenges:

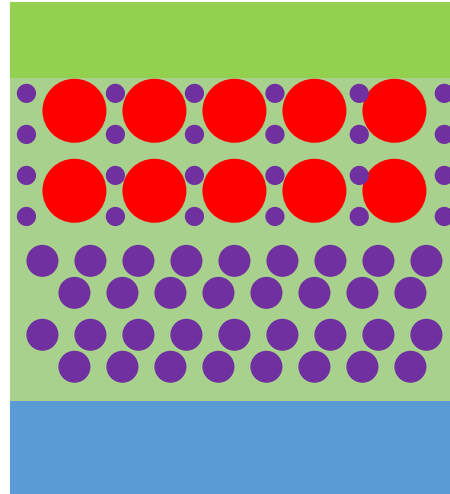
- Working at 70-90 °C
(low conductivity at RT)
- Low specific energy (~100 Wh/kg)
- PEO is not stable with 4 V cathode

Potential Solutions:

- Interfacial stabilization
- New polymer systems

Solid State Batteries with Composite Electrolytes

Al
NMC
Cathode
Ceramic/
polymer
electrolyte
Lithium



Advantages:

- Easy to process, ultrathin (10-20 μm)
- Enhanced ionic conductivity
- Ceramic fillers suppresses lithium dendrite
- Polymer wets electrode/electrolyte interface.
- A balance between performance & thermal stability

Oxide/PEO (Yang)



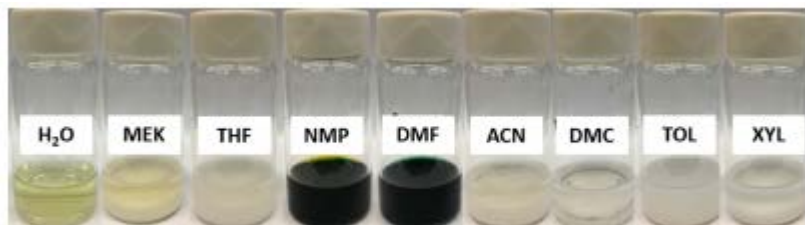
Sulfide/polymer (Chen and Meng)



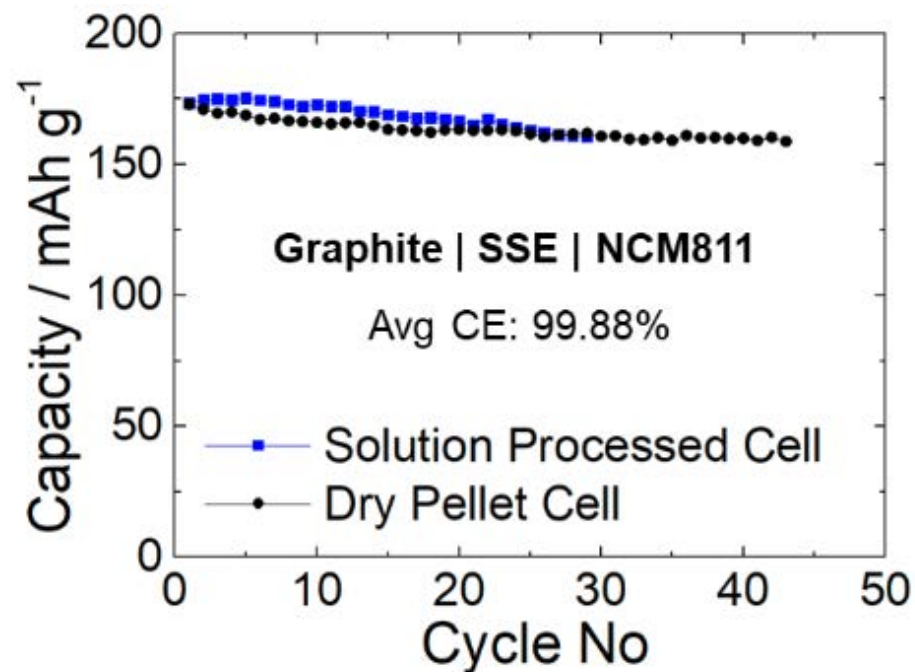
Oxide/Separator (Welion Inc)



Polymer/Sulfide Composite Electrolyte

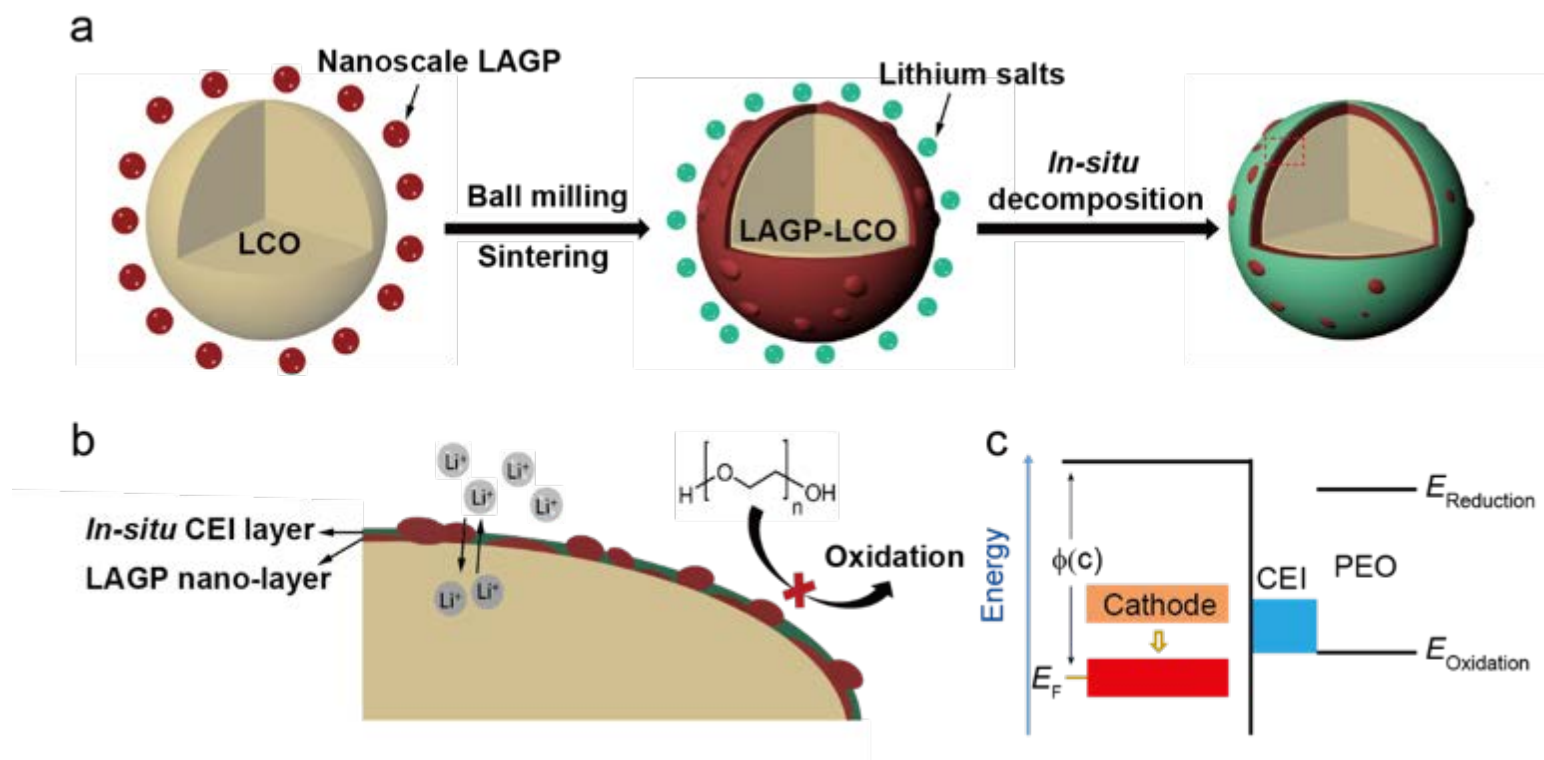
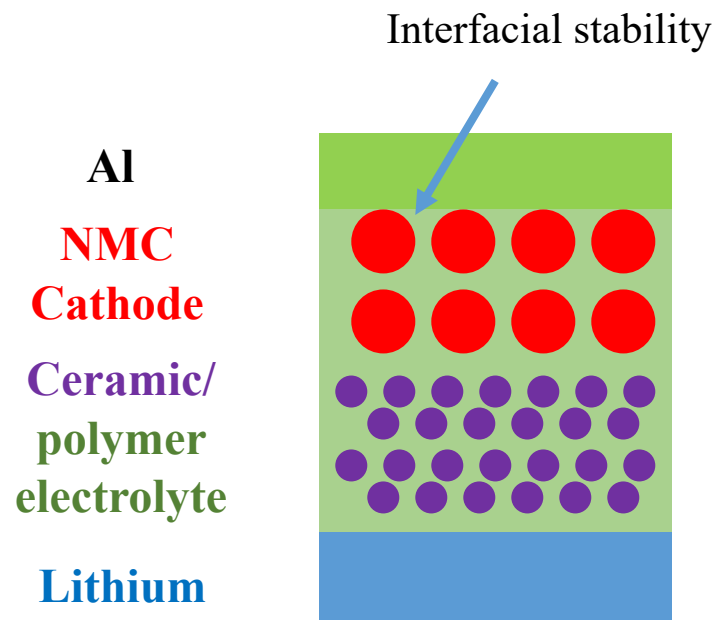


5 % Polymer/ 95% $\text{Li}_7\text{P}_3\text{S}_{11}$ electrolyte



Courtesy by
Zheng Chen

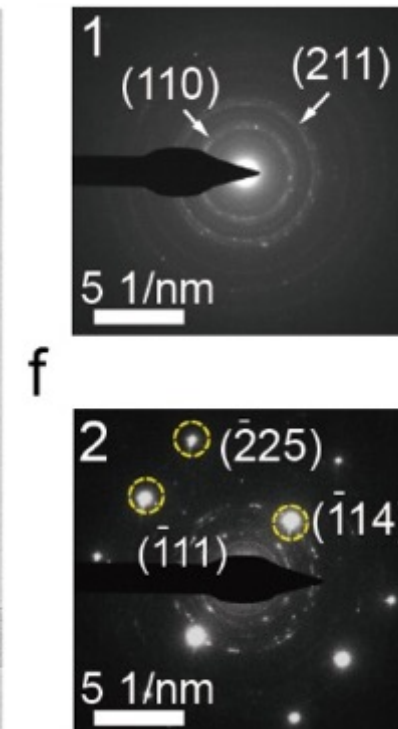
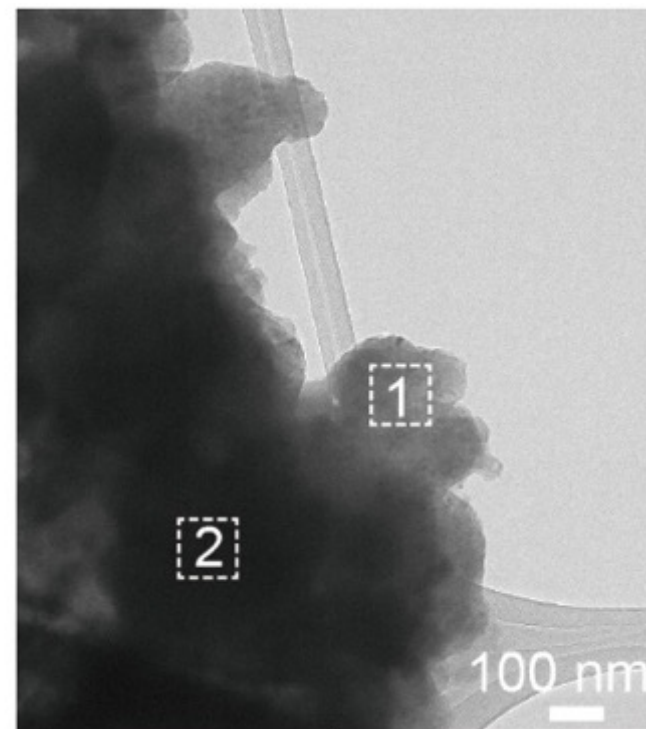
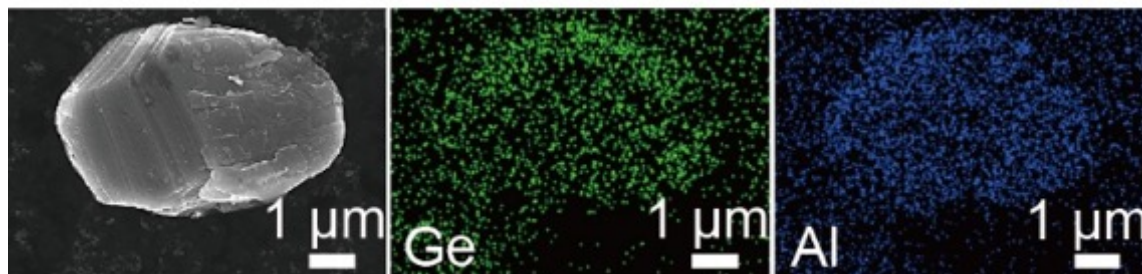
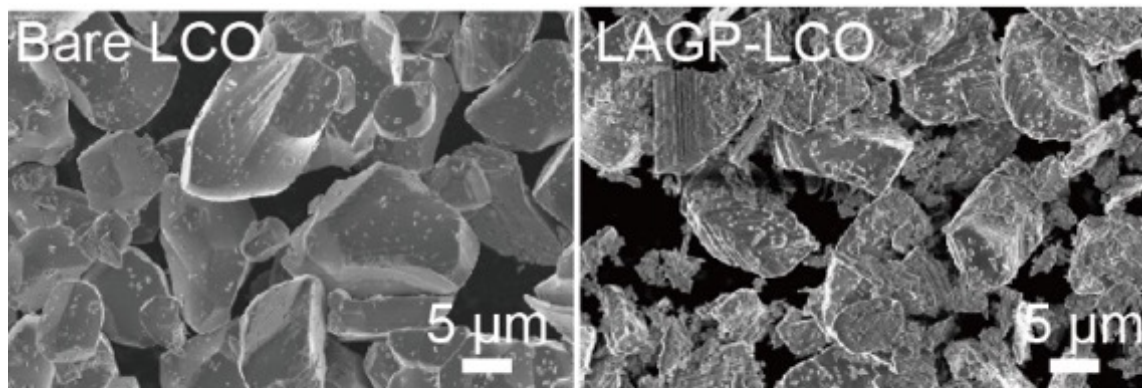
Double Layer Coating to Stabilize 4 V cathode/PEO interface



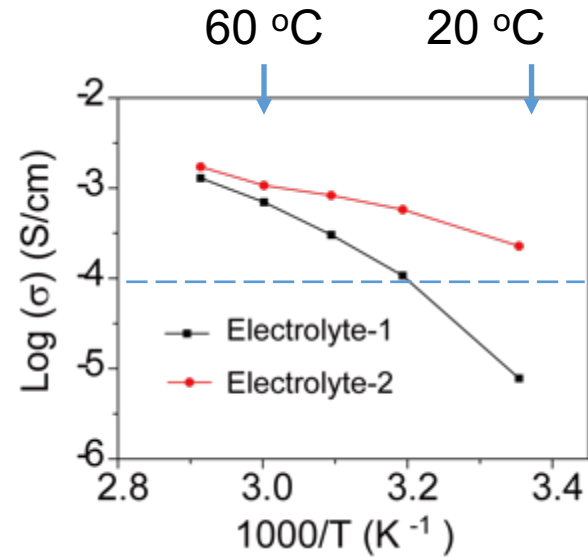
LAGP: $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}\text{P}_3\text{O}_{12}$

LAGP suppresses the decomposition of PEO,

The weak points are further protected by salt decomposition (e.g. LiDFOB, LiBOB)



Granular LAGP coating is formed on LCO by ball milling



Liquid electrolyte
1 M LiPF₆ in ECDEC

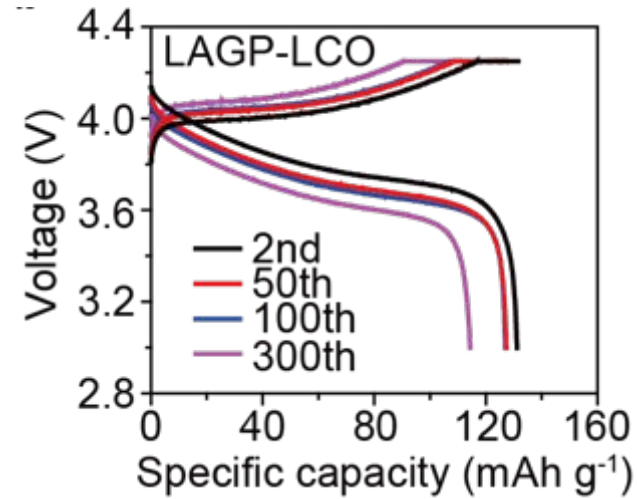
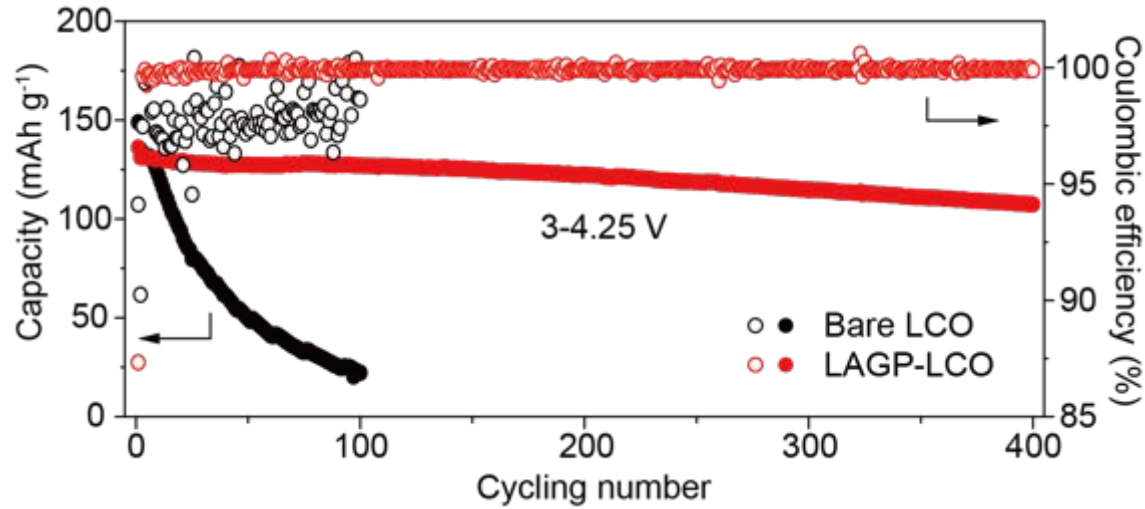
Electrolyte 1: 0.5 M Tri-salt in PEO-10K + PEG500 (1:1)

Electrolyte 2: Electrolyte 1 + LLZO

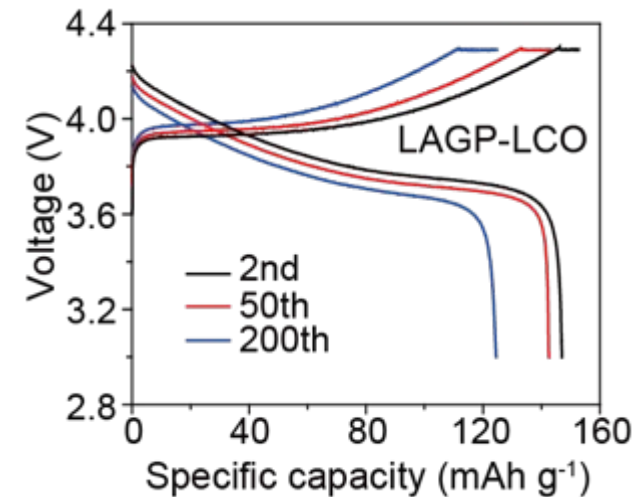
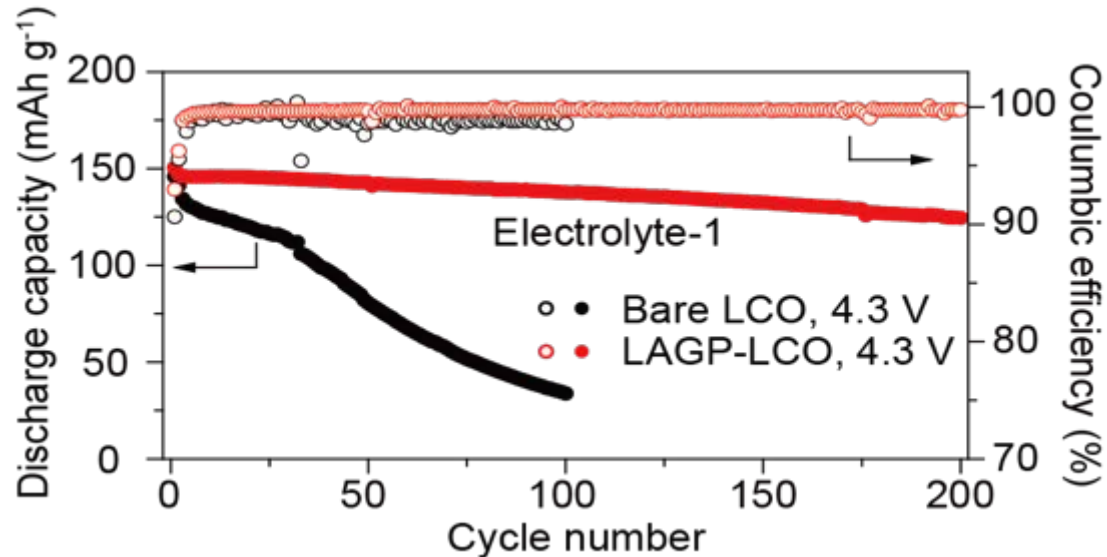
Tri-salt = LiTFSI+LiBOB+LiPF₆

Ignition Test

Cycling performance of LiCoO₂/PEO/Li cell at 60 °C

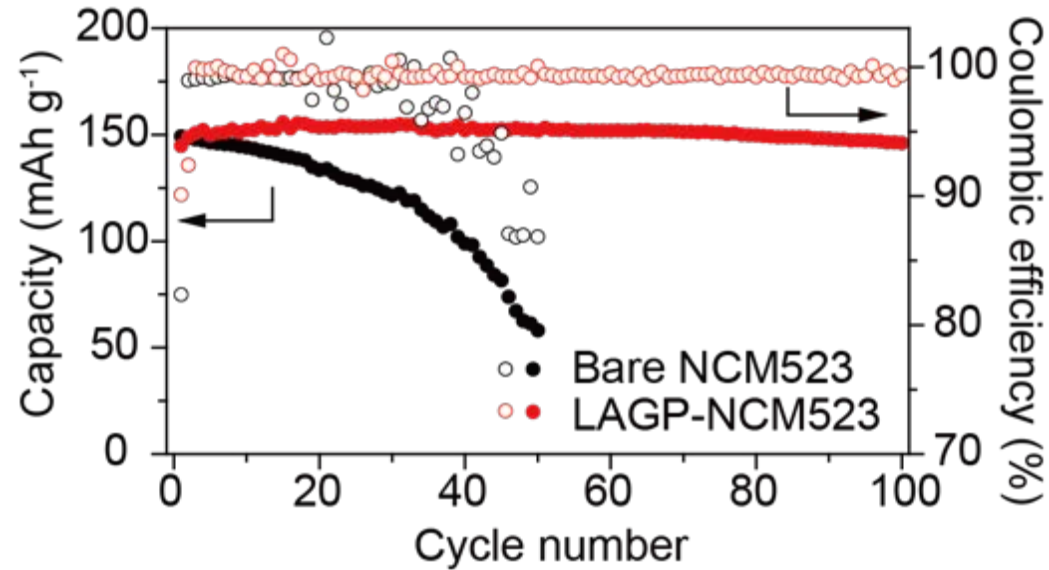


1 mAh/cm² 3.0 - 4.25 V,
82% / 400 cycles,
CE is ~99.9%
C: C/3, DC: C/2



1 mAh/cm² 3.0 - 4.3 V,
85% / 200 cycles,
CE is ~99.8%
C: C/3, DC: C/2

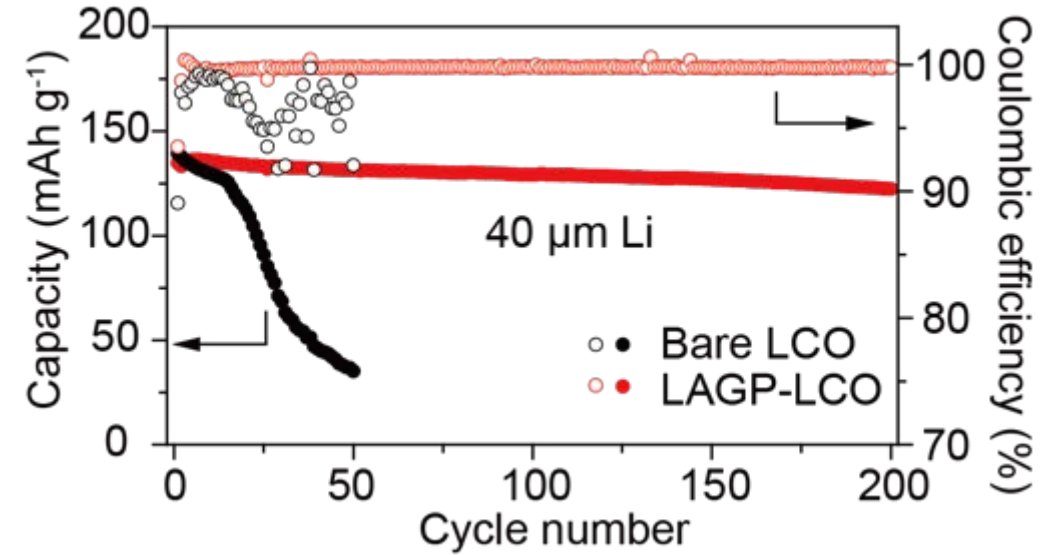
Performance with NCM and thin lithium



NCM523/Li (3-4.25 V)

1 mAh/cm² 0.3 C, 60 °C

93.8%/100 cycles

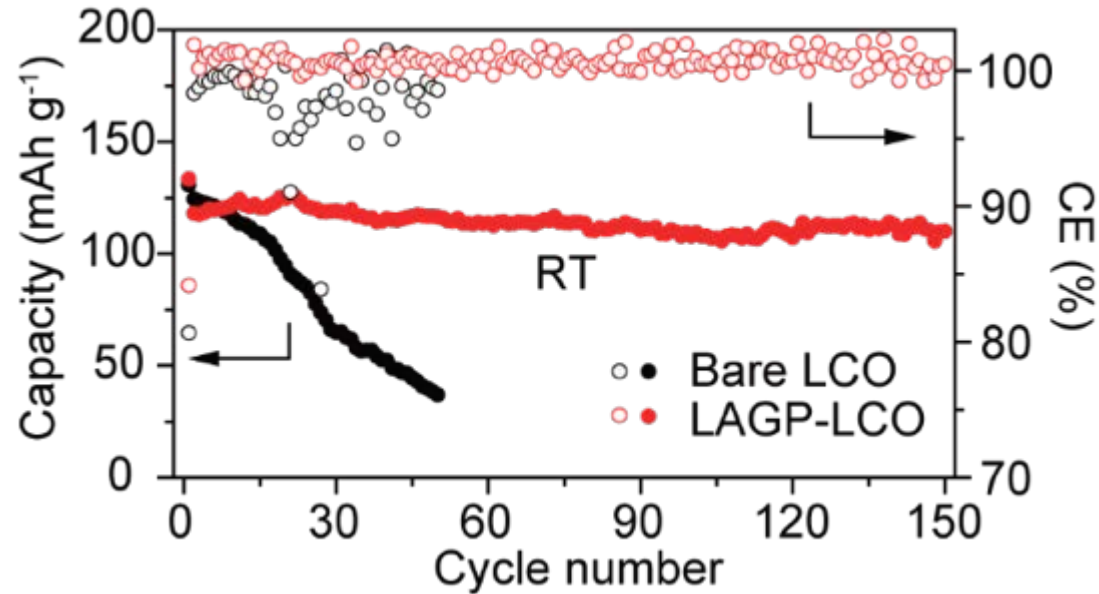


LCO/Li

1 mAh/cm² 0.3 C, 60 °C

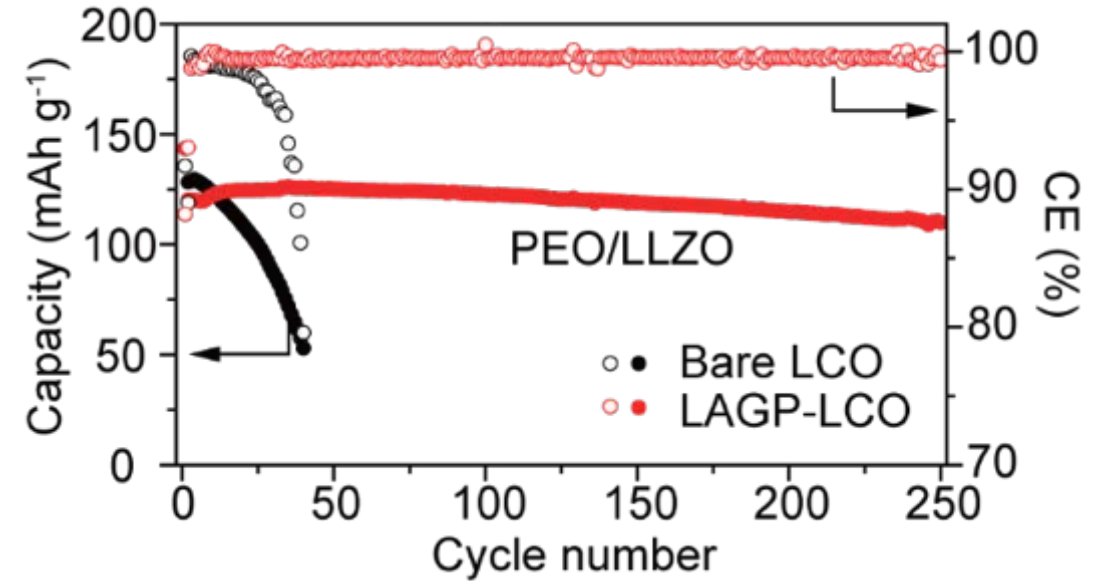
88%/200 cycles

Performance at RT and Composite Electrolyte



RT, 0.5 mAh/cm²
0.2 C, **90.8%**/100 cycles

25 wt% EC/PC plasticizer

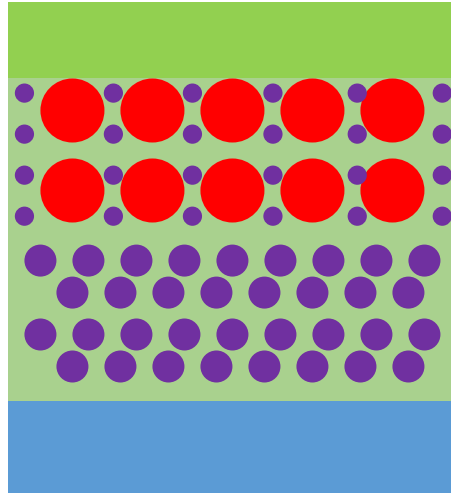


60 °C, 1 mAh/cm²
0.5 C, **87.2%**/250 cycles

PEO/LLZO electrolyte

Summary

Al
NMC
Cathode
Ceramic/
polymer
electrolyte
Lithium



- Scalability, interfacial stability/impedance, conductivity are critical factors to evaluate solid state batteries
- Composite electrolytes could combine advantages of both ceramic and polymer electrolytes
- PEO/4 V cathode is a possible combination.

Acknowledgement



Collaborator

Xi Chen (Columbia)

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ui Xiao (BNL)

anari Waluyo (BNL)

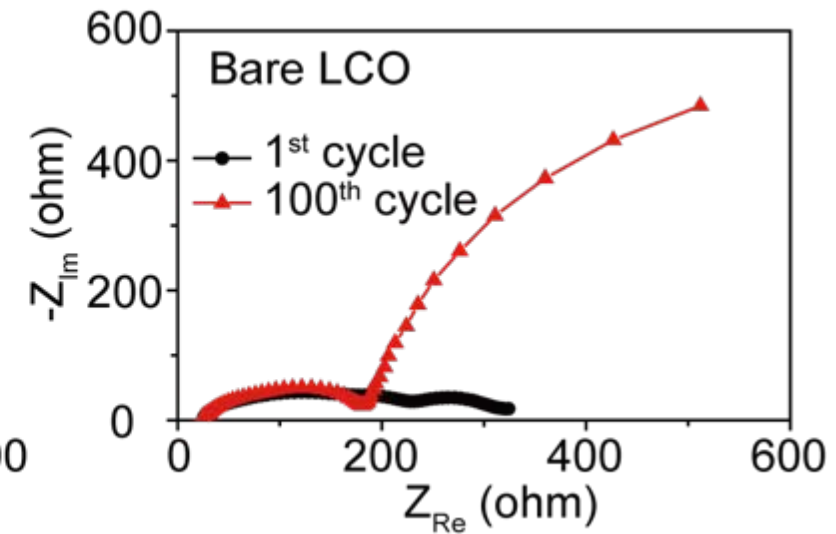
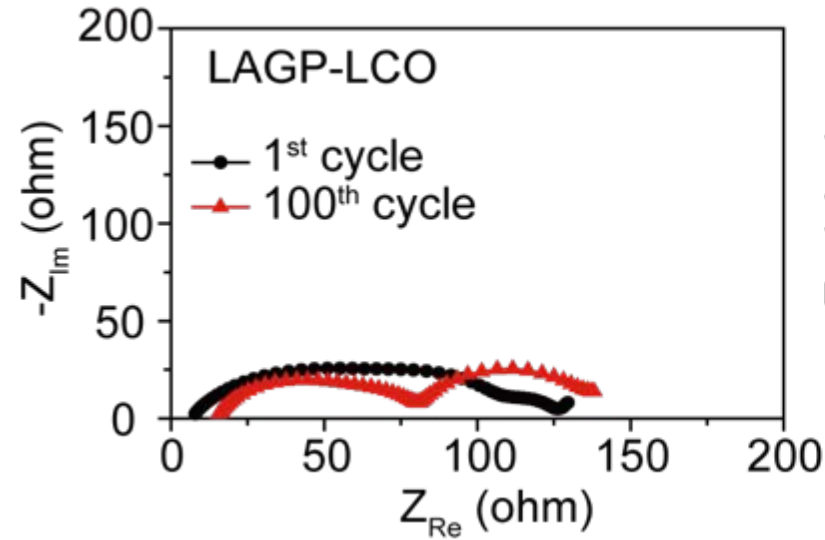
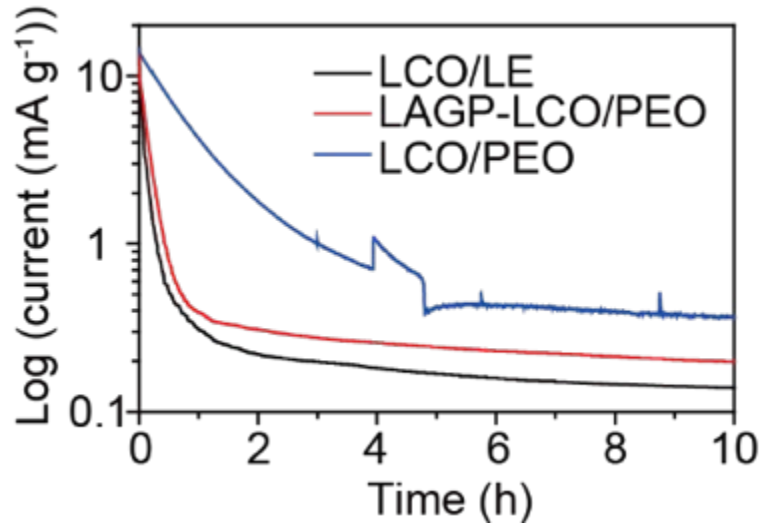
n Lin (BNL)

Xiaoqing Yang (BNL)



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Understanding the Protection Mechanisms



- Leakage current in LAGP-LCO approaches carbonate electrolyte (LE) when held at 4.3 V
- EIS shows that the degradation arises from the ion diffusion in the cathode.